University of Mumbai Examination 2020 under cluster ---- (Lead college Short name)

Program: Civil Engineering

Curriculum Scheme: Revised 2016

Examination: Third Year Semester V

Course Code and Course Name: CE-C 504- Environmental Engineering 1

Time: 1 hour Max. Marks: 50

Note to the students: - All Questions are compulsory and carry equal marks.

Q1.	The cost of water treatment is usually highest for which of the following
	sources?
Option A:	Groundwater
Option B:	River water
Option C:	Lake water
Option D:	Seawater
Q2.	The most common location for providing screens in water supply projects is:
Option A:	Entry point at water intake
Option B:	Before sedimentation unit at water treatment plant
Option C:	First unit at water treatment plant
Option D:	After aeration unit at water treatment plant
Q3.	If the level of source of water is higher than that of the place of consumption,
	the system adopted for supplying water is
Option A:	Pumping system
Option B:	Gravity system
Option C:	Combined pumping and gravity system
Option D:	Grid iron system
Q4.	Firefighting demand for a city/town is usually estimated, based on:
Option A:	Population
Option B:	Area of the city
Option C:	Tropical areas
Option D:	LPCD rate
Q5.	The process of nutrient enrichment is termed as,
Option A:	Limiting nutrients
Option B:	Eutrophication
Option C:	Enrichment
Option D:	Schistosomiasis

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Q6.	Which of the following is not a water borne disease?
Option A:	Scabies
Option B:	Typhoid
Option C:	Cholera
Option D:	Hepatitis
Q7.	All of the following are considered toxic metal pollutants EXCEPT
Option A:	Chromium
Option B:	Cadmium
Option C:	Potassium
Option D:	Mercury
Q8.	In a rapid sand filter
Option A:	Raw water from the source is supplied
Option B:	disinfected raw water is supplied
Option C:	Raw water passed through coagulation tank is supplied
Option D:	any one of the above
00	If the population of the city is 2 lacs and average water consumption is 240
Q9.	lpcd, then the water requirement of the city in million liters is:
Option A:	108 mld
Option B:	72 mld
Option C:	48 mld
Option D:	40 mld
Q10.	Which of the following ions are responsible for carbonate hardness?
Option A:	Carbonates & bicarbonates of calcium & magnesium
Option B:	Bicarbonates of sodium & potassium
Option C:	Carbonates of calcium & magnesium
Option D:	Dissolved carbon dioxide
Q11.	Pressure filters are not suited for
Option A:	swimming pools
Option B:	railway stations
Option C:	individual industries
Option D:	Public water supplies
<u> </u>	Dartiolog of cize around 1 migron (10:6) size are best removed by
	Particles of size around 1 micron (10 ⁻⁶) size are best removed by
Option A:	Plain sedimentation
Option A: Option B:	Plain sedimentation Filtration
Q12. Option A: Option B: Option C: Option D:	Plain sedimentation

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Q20.	Which components are responsible for temporary hardness in water?
Option A:	Sulfates, Chlorides
Option B:	Sulfates, Chlorides, nitrates
Option C:	Carbonates and bicarbonates
Option D:	Sulfates and carbonates
•	
Q21.	Disinfection of water helps in,
Option A:	Removing turbidity
Option B:	Removing hardness
Option C:	Killing pathogenic bacteria
Option D:	Complete sterilization
<u> </u>	
Q22.	The time period for which the water is stored in a sedimentation tank is
Ontion A.	called, Time of flow
Option A:	
Option B:	Settling velocity Detention time
Option C:	
Option D:	Settling time
Q23.	Calculate the quantity of alum required in kg to treat 10 mld water at the treatment plant where the alum dose required is 12 mg/lit.
Option A:	120kg
Option B:	180 kg
Option C:	100 kg
Option D:	50kg
Q24.	Which of these membrane processes are likely to have highest degree of rejection for water contaminants?
Option A:	Micro filtration
Option B:	Nano filtration
Option C:	Ultra-Filtration Ultra-Filtration
Option D:	Reverse Osmosis
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Q25.	A chlorination unit treating 26 MLD water feeds 32 kg chlorine daily. If the chlorine demand of water is 1 mg/L, the residual chlorine levels in treated water would be:
Option A:	0.77 mg/L
Option B:	0.23 mg/L
	0.22 //
Option C:	0.32 mg/L